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NEWS	2	DEC	01	ChemPort single article sales feature unavailable
NEWS	3	JAN	06	The retention policy for unread STNmail messages
NEWS	4	JAN	07	will change in 2009 for STN-Columbus and STN-Tokyo WPIDS, WPINDEX, and WPIX enhanced Japanese Patent
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NEWS	7	FEB	06	Patent sequence location (PSL) data added to USGENE
NEWS	8	FEB	10	COMPENDEX reloaded and enhanced
NEWS	9	FEB	11	WTEXTILES reloaded and enhanced
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NEWS	11	FEB	19	Increase the precision of your patent queries use terms from the IPC Thesaurus, Version 2009.01
NEWS	12	FEB	23	Several formats for image display and print options discontinued in USPATFULL and USPAT2
NEWS	13	FEB	23	MEDLINE now offers more precise author group fields and 2009 MeSH terms
NEWS	14	FEB	23	TOXCENTER updates mirror those of MEDLINE - more precise author group fields and 2009 MeSH terms
NEWS	15	FEB	23	Three million new patent records blast AEROSPACE into STN patent clusters
NEWS	16	FEB	25	USGENE enhanced with patent family and legal status display data from INPADOCDB
NEWS	17	MAR	06	INPADOCDB and INPAFAMDB enhanced with new display formats
NEWS	18	MAR	11	EPFULL backfile enhanced with additional full-text
NEWS	10	MAR	11	applications and grants ESBIOBASE reloaded and enhanced
NEWS		MAR		CAS databases on STN enhanced with new super role
CMTNI	20	MAN	20	for nanomaterial substances
NEWS	21	MAR	23	CA/CAplus enhanced with more than 250,000 patent equivalents from China
NEWS	22	MAR	30	IMSPATENTS reloaded and enhanced
NEWS	23	APR		CAS coverage of exemplified prophetic substances enhanced
NEWS	24	APR	07	STN is raising the limits on saved answers
NEWS		APR	24	CA/CAplus now has more comprehensive patent assignee

information

NEWS 26 APR 26 USPATFULL and USPAT2 enhanced with patent assignment/reassignment information

NEWS 27 APR 28 CAS patent authority coverage expanded

NEWS 28 APR 28 ENCOMPLIT/ENCOMPLIT2 search fields enhanced

NEWS 29 APR 28 Limits doubled for structure searching in CAS REGISTRY

NEWS EXPRESS JUNE 27 08 CURRENT WINDOWS VERSION IS V8.3, AND CURRENT DISCOVER FILE IS DATED 23 JUNE 2008.

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=> s prepar? (bisphenol (w) a)

MISSING OPERATOR 'PREPAR? (BISPHENOL'

The search profile that was entered contains terms or nested terms that are not separated by a logical operator.

=> s prepare? (s) (bisphenol (w) a)

148790 PREPARE?

147140 PREP

2499 PREPS

149408 PREP

(PREP OR PREPS)

2266330 PREPD

3 PREPDS

2266332 PREPD

(PREPD OR PREPDS)

2459863 PREPARE?

(PREPARE? OR PREP OR PREPD)

81242 BISPHENOL

5064 BISPHENOLS

82769 BISPHENOL

(BISPHENOL OR BISPHENOLS)

23280567 A

L1 8851 PREPARE? (S) (BISPHENOL (W) A)

=> s l1 (L) rectification

19023 RECTIFICATION

116 RECTIFICATIONS

19090 RECTIFICATION

(RECTIFICATION OR RECTIFICATIONS)

L2 2 L1 (L) RECTIFICATION

=> d 12 1-2 ibib abs

L2 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2005:41403 CAPLUS

DOCUMENT NUMBER: 142:375820

TITLE: New sideline extraction process for catalytic

rectification

INVENTOR(S): Qiu, Zhaorong; Wang, Cheli; Cheng, Minlian; Ye, Qing;

Yang, Jihe

PATENT ASSIGNEE(S): China Petrochemical Co., Ltd., Peop. Rep. China;

Jiangsu Petrochemical College

SOURCE: Faming Zhuanli Shenqing Gongkai Shuomingshu, 25 pp.

CODEN: CNXXEV

DOCUMENT TYPE: Patent LANGUAGE: Chinese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
CN 1478577	A	20040303	CN 2002-142233	20020827

CN 1247289 C 20060329

PRIORITY APPLN. INFO.: CN 2002-142233 20020827 The sideline extraction method for drawing the product and/or byproduct out during catalytic rectification by mounting an extractor mounted on the middle of the reaction region of the catalytic rectification tower is presented. The systems used include a solid-liquid system, a liquid-liquid system or its layered alternative, or a liquid-gas system. The liquid in the solid-liquid system may be separated by gravity separation method or filtration and fed back to the reaction region. The liquid-liquid system may be separated by membrane filtration, rectification, extraction, adsorption, absorption, gas stripping, etc., and one kind of liquid in the liquid-liquid system may be fed back to the reaction region, while the layered liquid-liquid system may be separated by gravity separation The extractor for the liquid-liquid system is an internal liquid

separator and an external liquid separator. An internal cooling separator is mounted in the top of the catalytic rectification tower, and used to cool and sep. the gas phase in the rectification tower. The method may be used in esterification, transesterification, saponification, hydrolysis, alkylation, isomerization, amination, oxidation, etherification, etc. Tri-Bu citrate, isobutylene, and bisphenol A were prepared by using the sideline extraction process.

L2 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1988:168545 CAPLUS

DOCUMENT NUMBER: 108:168545

ORIGINAL REFERENCE NO.: 108:27719a,27722a

TITLE: Process for producing polycarbonates which do not

cause corrosion during molding

INVENTOR(S): Koqa, Shinichiro; Matsuno, Akira; Sakata, Katsuyuki;

Otani, Yoshiaki; Akihara, Isao

PATENT ASSIGNEE(S): Mitsubishi Chemical Industries Co., Ltd., Japan

SOURCE: Eur. Pat. Appl., 11 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.		DATE
EP 251586	A2	19880107	EP 1987-305423		19870618
EP 251586	A3	19890125			
EP 251586	В1	19920429			
R: DE, IT, NL					
JP 62297320	A	19871224	JP 1986-142164		19860618
JP 06076482	В	19940928			
JP 62297321	A	19871224	JP 1986-142165		19860618
JP 06076483	В	19940928			
JP 63090536	A	19880421	JP 1986-235829		19861003
JP 03020132	В	19910318			
BR 8703052	A	19880308	BR 1987-3052		19870617
US 4839458	A	19890613	US 1987-63001		19870617
PRIORITY APPLN. INFO.:			JP 1986-142164	А	19860618
			JP 1986-142165	А	19860618
			JP 1986-235829	А	19861003
70 01 1 1 1	1 1	1 1 00	17 4 4 7 7 1 1 1	7 1	

AB Polycarbonates, which have reduced CCl4 (I), which, upon molding, do not

yellow or cause mold die corrosion are prepared by polymerizing a hydroxydiaryl compound, COC12, and, optionally, a diamine compound or an acid chloride in a CH2Cl2 solvent, where the concentration of I present in the COCl2 and CH2Cl2 satisfies the equation A + 5B < 200 [A = I concentration (ppm) in COC12; B = I concentration (ppm) in CH2Cl2]. COCl2 (containing 500 ppm I) was purified by rectification in a Widmer rectification pipe at column temperature 8° and reflux ratio 1:1, producing COC12 containing 5 ppm I. A polycarbonate was prepared by treating a solution containing 15.3 L H2O, 1.03 kg NaOH, 3.0 g Na hydrosulfite, and 2790 g bisphenol A, into which 3.1 L of an aqueous 10 wt/volume % NaOH solution and 7.65 L CH2Cl2 (containing 5 ppm I) were added, and 1.38 kg purified COCl2 were blown in under stirring for 60 min, followed by addition of 66.7 g 4-tert-BuC6H4OH and 3.1 g Et3N. The mixture was stirred for 1 h during which polymerization place. The organic layer was washed with H2O, neutralized with 3.0 L 10% H3PO4, washed with H2O three times, and heated to distill off CH2Cl2, producing a powdery polycarbonate which, after drying and pelletization had yellowness index 6. After molding 200 sheets from the pellets (using carbon steel molding dies; 300°) no molding die corrosion was observed

=> d his

L1

took

(FILE 'HOME' ENTERED AT 15:19:34 ON 28 APR 2009)

FILE 'CAPLUS' ENTERED AT 15:20:03 ON 28 APR 2009 8851 S PREPARE? (S) (BISPHENOL (W) A)

L2 2 S L1 (L) RECTIFICATION

=> s l1 (l) (recycle (s) effluent)

33193 RECYCLE

2160 RECYCLES

34963 RECYCLE

(RECYCLE OR RECYCLES)

98794 EFFLUENT

51189 EFFLUENTS

132938 EFFLUENT

(EFFLUENT OR EFFLUENTS)

L3 0 L1 (L) (RECYCLE (S) EFFLUENT)

=> s 11 and (recycle (s) byproduct)

33193 RECYCLE

2160 RECYCLES

34963 RECYCLE

(RECYCLE OR RECYCLES)

40520 BYPRODUCT

33749 BYPRODUCTS

67357 BYPRODUCT

(BYPRODUCT OR BYPRODUCTS)

339 RECYCLE (S) BYPRODUCT

L4 1 L1 AND (RECYCLE (S) BYPRODUCT)

=> d 14 ibib abs

L4 ANSWER 1 OF 1 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2001:355038 CAPLUS

DOCUMENT NUMBER: 134:340822

TITLE: Preparation and crystallization process for the

manufacture of high-purity bisphenol A

INVENTOR(S): Heydenreich, Frieder; Prein, Michael; Boediger,

Michael; Neumann, Rainer

PATENT ASSIGNEE(S): Bayer A.-G., Germany SOURCE: Ger. Offen., 6 pp.

CODEN: GWXXBX

DOCUMENT TYPE: Patent LANGUAGE: German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PA	TENT				KIND DATE				APPLICATION NO.						DATE			
		4786			A1 20010517			DE 1999-19954786										
WC	2001	2001036358			A1 20010525			WO 2000-EP10827						20001103				
	W:	ΑE,	AG,	AL,	ΑM,	ΑT,	ΑU,	ΑZ,	BA,	BE	3, BG,	BR,	BY,	ΒZ,	CA,	CH,	CN,	
		CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EE,	ES	S, FI,	GB,	GD,	GE,	GH,	GM,	HR,	
		HU,	ID,	IL,	IN,	IS,	JP,	KΕ,	KG,	KF	P, KR,	KΖ,	LC,	LK,	LR,	LS,	LT,	
		LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX	K, MZ,	NO,	NΖ,	PL,	PT,	RO,	RU,	
		SD,	SE,	SG,	SI,	SK,	SL,	ТJ,	TM,	TF	R, TT,	TZ,	UA,	UG,	US,	UZ,	VN,	
		YU,	ZA,	ZW														
	RW:	GH,	GM,	KΕ,	LS,	MW,	MZ,	SD,	SL,	SZ	Z, TZ,	UG,	ZW,	ΑT,	BE,	CH,	CY,	
		DE,	DK,	ES,	FΙ,	FR,	GB,	GR,	ΙE,	ΙΊ	Γ, LU,	MC,	NL,	PT,	SE,	TR,	BF,	
		,	,	,					,		L, MR,	,	,	,				
		2001010277									AU 2001-10277							
		2000015555							BR 2000-15555									
EP	1232	1232134				A1 20020821					EP 2000-971412					20001103		
EP	1232	2134			B1 20040922													
	R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GF	R, IT,	LI,	LU,	NL,	SE,	MC,	PT,	
		ΙE,	SI,	LT,	LV,	FΙ,	RO,				I, TR							
JP	2003	35239	50		T		20030812 JP 2001-538314							0001				
ΑT	2769	90			Τ			AT 2000-971412							0001			
CN	1189	438			С	T 20030812 T 20041015 C 20050216			CN 2000-815651							0001		
ES	2228	3621			Т3	T3 20050416			ES 2000-971412						0001			
	₹ 226323				В	B 20050111										0001		
	IN 2002MN00520						2006				2002-					0020		
	MX 2002004812														0020	-		
US	6710	211			В1		2004	0323		US	2002-	1299	44		2	0020		
RIORIT	Y APE	LN.	INFO	.:							1999-							
										WO	2000-	EP10	827		W 2	0001	103	

AB Highly pure bisphenol A, prepared by the condensation of phenol with acetone in the presence of an acidic sulfonated polystyrene resin cation exchanger catalyst, is purified by:

(A) a primary crystallization in the form of a continuous or discontinuous layer

crystn; (B) subjecting it to an optional distillation or crystallization; and (C)

removing water, acetone, and phenol from the byproduct stream for recycle to the initial reactor. A process flow diagram is presented.

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	FILE	'CAPLU	JS	' E1	NTERI	ED AT	15:2	0:03	3 ON	28	APR	2009
L1		8851	S	PRE	EPARI	E? (S)	(BI	SPHE	ENOL	(W)	) A)	
L2		2	S	L1	(L)	RECTI	FICA	OITA	1			
L3		0	S	L1	(L)	(RECY	CLE	(S)	EFFI	LUEI	(TV	
L4		1	S	L1	AND	(RECY	CLE	(S)	BYPE	RODI	JCT)	

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